

Effect of combined doping ($Y_2O_3+Fe_2O_3$) on structural peculiarities of nanodispersed ZrO_2

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The most effective stabilizer for zirconium oxide is yttrium oxide. However, the structure of Y- ZrO_2 degraded at low temperature. Partial substitution of Fe^{3+} for Y^{3+} in system $Y_2O_3-ZrO_2$ decreases both the crystallization and sintering temperature of zirconia ceramic. It is known that the content of monoclinic (M), tetragonal (T) and cubic (C) polymorphs determines the properties of ZrO_2 .

The aim of present work is the investigation of structural peculiarities (polymorphs, positions of atoms, site occupancies, local environment of Fe^{3+}) of zirconium oxide stabilized by combined dopant (Y_2O_3 and Fe_2O_3) depending on chemical composition ($(1-x)ZrO_2 \cdot xY_2O_3 \cdot yFe_2O_3$, where $x+y=0.03-0.08$), synthesis conditions (coprecipitation of hydroxides or successive precipitation of hydroxides) and heat treatment (970-1570 K).

It has been shown that solubility of iron in Y- ZrO_2 increases with yttrium content. Iron dissolves completely in Y- ZrO_2 at $Y/Fe \geq 2$. Increasing Y/Fe ratio in ZrO_2 doped with the same total amount of doping oxides stabilizes the structure and inhibits low-temperature degradation. Increasing the total amount of doping oxides extends the temperature range of existence of C and C+T polymorphs of ZrO_2 . Mössbauer spectra of fully stabilized tetragonal Y-Fe- ZrO_2 showed that distribution of Fe^{3+} ions has a cluster topology. Two nonequivalent sites of Fe^{3+} with octahedral coordination in coprecipitated samples and three nonequivalent sites of Fe^{3+} with octa-, penta- and tetrahedral coordination in successively precipitated samples have been identified. Decrease in coordination number of iron ions in comparison with that of host cations in Y- ZrO_2 stabilizes the structure and inhibits its degradation due to increase in Me-O binding energy. It has been shown that precipitated ZrO_2 powders contain nanoparticles with grain size of 10-20 nm. Successively precipitated powders, in contrast to coprecipitated ones, consist of soft easy-breaking aggregates and do not require additional grinding.