

Sol-gel preparation and luminescent properties of CeO₂:Ln (Ln = Eu³⁺ and Sm³⁺) thin films

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Abstract

Recent progress in ultraviolet light emitting diodes (UVLEDs) has opened possibility of designing new opto-electronic devices. UV-pumped phosphors are one of the key materials for such devices. Especially, luminescence from rare earth shows sharp spectrum, which is effective phosphors in display device, solid-state laser and etc. In this work, new transparent CeO₂:Ln (Ln = Eu³⁺ or Sm³⁺) thin-film phosphors have been fabricated by a sol-gel method. Homogeneous precursor solutions, with Ln concentrations varied between 0.05 and 100 at.%, were obtained by dissolving a mixture of pertinent metal acetates and trifluoroacetic acid in ethanol. The solutions were deposited on silica glass substrates by spin-coating and then heated at 700°C in air. Fluorite-type crystalline phases were formed in all the films, although their chemical compositions seemed to change from CeO₂:Ln to LnOF depending on the Ln concentration. CeO₂:Ln films with the Ln concentrations of 1 at.% showed the strongest photoluminescence (PL) upon ultraviolet (UV) light excitation. At higher Ln concentrations, the PL intensity decreased because of the nonradiative energy transfer between Ln ions as luminescent centers. For Eu³⁺ ions doped in the CeO₂ lattice with inversion symmetry, an orangish red emission due to a magnetic-dipole ⁵D₀ → ⁷F₁ transition (591 nm) was dominant without other strong ⁵D₀-related emissions when excited at a wavelength of 340 nm. In contrast, much stronger orange emissions (573 nm) due to ⁴G_{5/2} → ⁶H_{5/2} were observed for Sm³⁺ ions. Apparently, the Eu³⁺ and the Sm³⁺ ions had different efficiencies of the excitation through the energy transfer from the “Ce⁴⁺-O²⁻” charge transfer state induced by the UV irradiation.